CB Reinforced Polymer Nanocomposites: A Research Review

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Abstract

Carbon Black reinforced polymer nanocomposites (CBRPNs) have become a focal point in materials science due to their improved mechanical, electrical, and thermal properties. This review provides a thorough summary of the synthesis, characteristics, and applications of CBRPNs. It examines different synthesis techniques, such as in-situ polymerization, melt mixing, and solution blending. The review delves into the influence of carbon black (CB) on the mechanical strength, electrical conductivity, and thermal stability of polymer matrices. Furthermore, it looks into the various uses of these nanocomposites in industries like automotive, aerospace, electronics, and biomedical fields. The review concludes with perspectives on future research directions and challenges in the advancement and commercialization of CBRPNs.

Keywords

Carbon Black, Nanocomposites, Rubber, Polyethylene, Thermosets

I. INTRODUCTION

Polymer nanocomposites represent a groundbreaking frontier in materials science, merging the advantageous properties of polymers with the unique attributes of nanoscale fillers. These advanced materials are composed of a polymer matrix containing nanoparticles, which measure between 1 and 100 nanometers in size. The incorporation of these nanoparticles into the polymer matrix results in a synergistic enhancement of material properties, far surpassing the capabilities of conventional composites and unfilled polymers. The versatility of polymer nanocomposites stems from their ability to significantly improve mechanical strength, thermal stability, electrical conductivity, and barrier properties, making them indispensable in a variety of high-performance applications. From lightweight automotive and aerospace components to advanced electronics, packaging solutions, and medical devices, polymer nanocomposites are revolutionizing industries with their superior performance characteristics. This introduction delves into the fascinating world of polymer nanocomposites, exploring their composition, properties, applications, and the innovative processing techniques used to create them. As research and development expand the capabilities of these materials, polymer nanocomposites are leading the way in nextgeneration material solutions, ready to propel technological advancements and sustainable innovations across various sectors.

The advent of polymer nanocomposites has transformed material science by providing superior properties that traditional composites cannot achieve. Among the various nanofillers, carbon black (CB) stands out due to its distinctive attributes, including a high surface area, electrical conductivity, and cost-effectiveness. When integrated into polymer matrices, CB can greatly improve mechanical, electrical, and thermal properties, making it ideal for numerous applications.

II. SYNTHESIS METHODS

The synthesis of CB reinforced polymer nanocomposites can be broadly categorized into three methods: in-situ polymerization, melt mixing, and solution blending.

A. In-situ Polymerization

In this method, monomers are polymerized in the presence of CB particles. This method guarantees a uniform dispersion of carbon black within the polymer matrix, leading to improved interfacial interaction. It results in improved properties and their stability due to the uniform dispersion and strong interaction between the polymer matrix and CB nanoparticles. Techniques such as emulsion polymerization, suspension polymerization, and bulk polymerization are commonly used. The key steps and elements involved in the in situ polymerization process for creating CB reinforced polymer nanocomposites is as shown in Figure 1.

Figure 1 In-situ Polymerization method

B. Melt Mixing

This method involves melting the polymer and mixing it with CB particles at high temperatures. The process is straightforward and suitable for thermoplastic polymers. However, achieving uniform dispersion can be challenging due to the high viscosity of the molten polymer. In this method CB nanoparticles are heated above their melting point in an extruder or internal mixer to create a molten polymer. These are gradually added to the molten polymer. The blend is thoroughly mixed to achieve an even distribution of carbon black throughout the polymer matrix. The homogeneous mixture is extruded or molded into the desired shape. The material is cooled and solidified to form the final nanocomposite. A brief overview is as shown in Figure 2.

Figure 2 Melt Mixing Method

C. Solution Blending

In this technique, both the polymer and carbon black are dissolved or dispersed in a common solvent. After the solvent evaporates, a well-dispersed nanocomposite remains, as illustrated in Figure 3. This method works well for polymers that readily dissolve in solvents, although it is less environmentally friendly due to the use of organic solvents. It is particularly useful for creating uniform nanocomposites with precise control over the dispersion of fillers.

Figure 3 Solution Blending Method

III. TYPES OF CB NANOCOMPOSITES

A. CB/Rubber Nanocomposites

The use of CB in rubber matrices has been extensively studied. These nanocomposites exhibit enhanced tensile strength, abrasion resistance, and fatigue life, making them ideal for automotive tires, conveyor belts, and other rubber products [1]. The reinforcing effect of CB in rubber is well-documented and forms the basis for its application in other polymer systems. Elastomers, including natural rubber and synthetic rubbers like styrene-butadiene rubber (SBR) and ethylene propylene diene monomer (EPDM), when reinforced with CB, show remarkable improvements in elasticity, resilience, and mechanical strength [2]. These nanocomposites are essential in applications requiring high elasticity and durability. In

Automotive Industry the CBRPNs are used in tire manufacturing to improve wear resistance and reduce rolling resistance. They are also utilized in various automotive parts for enhanced durability and weight reduction. The lightweight and high-strength characteristics of CBRPNs make them ideal for aerospace applications, where material performance and weight are critical [3]. The enhanced electrical conductivity of CBRPNs makes them suitable for electronic components, including conductive films, sensors, and printed circuit boards. Due to their biocompatibility and improved mechanical properties, CBRPNs are explored for use in medical devices, prosthetics, and drug delivery systems [4].

Carbon black (CB) reinforced natural rubber (NR) nanocomposites mark a significant leap forward in rubber technology, providing superior mechanical, thermal, electrical, and barrier properties compared to traditional rubber materials. This blend capitalizes on the reinforcing benefits of CB alongside the natural elasticity and resilience of rubber, making them suitable for numerous applications across various industries. Derived from the latex sap of rubber trees (Hevea brasiliensis), natural rubber boasts excellent elasticity, resilience, and abrasion resistance, which are essential for applications demanding these characteristics [5]. However, pure natural rubber can lack sufficient strength and durability for demanding applications. CB is a form of elemental carbon produced by incomplete combustion or thermal decomposition of hydrocarbons. It is characterized by its high surface area, electrical conductivity, and reinforcing properties. In rubber applications, CB serves as a filler to enhance mechanical properties such as tensile strength, modulus, and abrasion resistance. The addition of CB enhances the tensile strength and modulus of natural rubber by acting as reinforcing agents, thereby improving the load-bearing capacity and stiffness of the rubber matrix. This reinforcement also enhances the abrasion resistance of natural rubber, making it more durable and suitable for applications involving wear and tear. CB forms a conductive network within the natural rubber matrix, imparting electrical conductivity beneficial for applications requiring static dissipation or EMI shielding [6]. While natural rubber itself has low thermal conductivity, the addition of CB can improve heat dissipation properties, although to a lesser extent compared to materials like thermoplastics or thermo sets. CB reduces gas permeability through natural rubber, enhancing gas barrier properties suitable for applications such as tire inner tubes and automotive seals. During manufacturing, CB is typically mixed with natural rubber using methods like Banbury mixing or two-roll milling to achieve uniform dispersion. The compounded mixture undergoes curing (vulcanization) with heat and sulfur or other curing agents to cross-link rubber molecules, enhancing mechanical properties and stability. CB-reinforced natural rubber finds extensive use in tire manufacturing due to its enhanced wear resistance, traction, and durability. It is also employed in seals, gaskets, conveyor belts, hoses, and tubing where flexibility, durability, and resistance to chemicals are crucial. In consumer goods and sports equipment, CBreinforced natural rubber is used in shoe soles, outsoles, balls, grips, and mats for enhanced abrasion resistance, comfort, elasticity, and durability. Industrial applications include vibration isolation mounts, pads, and seismic bearings where CB-reinforced natural rubber dampens vibrations, reduces noise transmission, and absorbs energy during seismic events. Natural rubber is biodegradable, contributing to its environmental sustainability compared to synthetic rubber alternatives. CB-reinforced natural rubber can also be recycled or reused in various applications, reducing environmental impact. Challenges include achieving uniform CB dispersion in the rubber matrix, which affects properties such as mechanical strength and conductivity. Additionally, cost considerations of CB and processing methods may influence the adoption of these nanocomposites in specific applications. Continued research focuses on optimizing CB loading, dispersion methods, and exploring alternative fillers to further enhance properties and reduce environmental impact. In conclusion, CB reinforced natural rubber nanocomposites offer a compelling combination of properties suitable for diverse industrial applications. Their enhanced mechanical strength, abrasion resistance, electrical conductivity, and barrier properties make them valuable materials in automotive, industrial, consumer goods, and construction sectors. Ongoing research and development efforts are expected to drive further innovations, improving performance, sustainability, and costeffectiveness of these nanocomposites in the future.

CB reinforcement offers significant enhancements to various key properties of styrene butadiene rubber (SBR) nanocomposites. Acting as a reinforcing filler, CB improves mechanical properties of SBR by fostering strong interactions with the SBR matrix, which in turn restrict polymer chain mobility and enhance resistance to deformation. Incorporating CB also enhances the tear resistance of SBR nanocomposites by acting as barriers that impede tear propagation, thereby increasing the energy required for tearing the material. Moreover, CB addition is known to enhance the abrasion resistance of rubber compounds, extending their durability, which is particularly advantageous in applications such as tire treads. CB also affects the dynamic properties of SBR, influencing aspects like damping behavior and rolling resistance, crucial in applications where energy dissipation and heat generation are significant [7]. Depending on CB type and concentration, SBR nanocomposites can exhibit improved electrical conductivity, a property highly valued in applications like sensors and electronic components. It's worth noting that the specific enhancements and their extent depend on factors such as CB type, concentration, processing techniques, and the presence of other additives. The specific surface area of CB particles significantly impacts their efficiency as reinforcement in SBR nanocomposites; higher specific surface areas, resulting from smaller CB particles, offer a larger total surface area per unit weight of filler [8]. This results in a more extensive interfacial area between the CB and the SBR matrix. The larger interfacial area allows for more interaction points between the CB particles and the SBR chains, resulting in stronger physical and chemical interactions, such as Van der Waals forces and hydrogen bonding. When stress is applied to the SBR nanocomposite, the larger interfacial area and stronger interactions facilitate more efficient stress transfer from the rubber matrix to the CB particles. This means the CB particles can effectively carry a larger portion of the load, preventing premature failure of the material. As a result of the improved stress transfer, SBR nanocomposites with higher surface area CB generally exhibit greater resistance to pulling forces before breaking, higher tear resistance, and increased wear and tear resistance from friction [9]. However, it is important to note that excessively high surface area CB can lead to processing difficulties due to increased viscosity and may not always translate to optimal reinforcement depending on the specific application and other compounding factors.

B. CB/Thermoplastic Nanocomposites

Thermoplastic polymers like polyethylene (PE), polypropylene (PP), and polystyrene (PS) shows significant improvements with the addition of carbon black (CB). These nanocomposites show enhanced mechanical properties, such as higher modulus and strength, along with better electrical conductivity, making them ideal for use in electronic applications and conductive packaging materials.

Polyethylene (PE) is widely utilized globally due to its excellent chemical resistance, ease of processing, and cost-effectiveness, finding extensive applications in packaging, construction, and consumer goods. However, its low mechanical strength and poor electrical conductivity limit its use in high-performance applications. Incorporating CB into PE matrices has emerged as a promising approach to enhance its properties, leading to the development of polyethylene/CB nanocomposites (PE/CB nanocomposites). The addition of CB significantly improves the mechanical properties of polyethylene, enhancing tensile strength, modulus, and impact resistance [10]. The effectiveness of these enhancements hinges on how well CB is dispersed and adheres to the polyethylene matrix. CB forms a conductive network within the polyethylene, greatly increasing electrical conductivity, which is advantageous for applications requiring antistatic materials and electromagnetic interference (EMI) shielding. PE/CB nanocomposites also demonstrate enhanced thermal stability and thermal conductivity compared to pure polyethylene, owing to CB's high thermal conductivity [11]. This property enables better heat dissipation, making these nanocomposites suitable for thermal management applications.

Polypropylene (PP) finds extensive applications in packaging, automotive components, and consumer goods. However, its relatively low mechanical strength and poor electrical conductivity limit its use in high-performance applications. The incorporation of CB into polypropylene matrices has been explored to overcome these limitations, resulting in the development of polypropylene/CB nanocomposites (PP/CB nanocomposites) [12]. The tensile strength of PP/CB nanocomposites is significantly higher than that of pure polypropylene. CB particles act as reinforcing agents, providing improved load transfer and stress distribution within the polymer matrix. There is an increase in the Young's modulus, indicating a stiffer material. The rigidity of CB particles contributes to the overall stiffness of the composite. Improved impact resistance is observed in PP/CB nanocomposites. The presence of CB helps in absorbing and dissipating impact energy more effectively than the neat polymer. Flexural strength is enhanced, making the material more resistant to bending forces. CB particles reinforce the polymer matrix, providing additional support and reducing deformation under stress [13]. A substantial increase in electrical conductivity is noted. CB forms a conductive network within the polymer matrix, allowing electrons to pass through the composite more efficiently. Improved antistatic properties help in preventing the buildup of static electricity. The conductive nature of CB dissipates static charges more effectively. Thermal stability is significantly improved, with higher degradation temperatures observed. CB particles enhance the thermal resistance of the polymer matrix by acting as heat barriers. Their increased thermal conductivity allows for better heat dissipation. CB has high thermal conductivity, which aids in the efficient transfer of heat through the composite. Improved flame retardancy is achieved with CB reinforcement, as the presence of CB can contribute to the formation of a protective char layer on the composite surface during combustion, which slows down the burning rate [14].The modification of melt flow properties can be customized based on the quantity and dispersion of CB. CB influences the viscosity and flow behavior of the molten polymer, improving processing characteristics. These enhancements in visco elastic properties contribute to better shape retention and resistance to deformation under mechanical stress. The interaction between CB particles and the polymer matrix enhances the resilience of the composite structure. Moreover, CB particles create a more convoluted pathway for gas molecules, thereby reducing gas permeability. This feature also extends to moisture barrier properties, where CB obstructs the passage of water vapor through the polymer matrix, enhancing moisture resistance. There is increased resistance to abrasion and wear. The reinforcing effect of CB particles provides a tougher surface that can withstand mechanical wear and tear. Improved resistance to UV degradation is also observed. CB can absorb UV radiation and protect the polymer matrix from photo-degradation [15]. Increased surface hardness provides better scratch resistance, as CB particles reinforce the surface layer of the composite, making it harder and more resistant to surface damage. Finally, improved surface finish and reduced surface defects are achieved. Proper dispersion of CB can enhance the overall appearance and uniformity of the composite surface.

Polystyrene (PS), widely favored for its ease of processing, clarity, and rigidity, exhibits enhanced properties when formulated into nanocomposites. Introducing CB into polystyrene significantly boosts its tensile strength and modulus [16]. Acting as reinforcing agents, CB particles enhance load distribution and facilitate stress transfer within the polymer matrix. Furthermore, the addition of CB enhances the impact resistance of polystyrene nanocomposites by absorbing and dissipating impact energy, thereby reducing crack propagation. One of the most notable improvements in CB/polystyrene nanocomposites is their electrical conductivity, as CB forms a conductive network within the polymer matrix, transforming insulating polystyrene into a conductive material essential for applications requiring antistatic properties and electromagnetic interference (EMI) shielding [17]. Additionally, the thermal stability of polystyrene is significantly heightened with CB addition, as CB particles mitigate polymer matrix degradation at elevated temperatures. Moreover, the nanocomposites' thermal conductivity increases due to CB's high thermal conductivity. This allows for better heat dissipation in applications where thermal management is critical. The incorporation of CB affects the viscosity and flow behavior of polystyrene, which can be advantageous during the processing and molding stages. The tailored rheological properties allow for better processing characteristics and improved control over the final product's shape and structure. CB/polystyrene nanocomposites exhibit improved barrier properties against gases and moisture. The presence of CB within the polymer matrix creates a more convoluted pathway for gas and moisture molecules, thereby reducing their permeability [18]. Incorporating CB enhances the abrasion and wear resistance of polystyrene nanocomposites by imparting a tougher surface that withstands mechanical wear and tear, thereby extending material lifespan in demanding applications. The improved electrical conductivity and electromagnetic interference (EMI) shielding capabilities of CB/polystyrene nanocomposites make them well-suited for electronic components, housings, antistatic packaging, conductive coatings, and enclosures for electronic devices. These nanocomposites also find applications in the automotive industry for various interior and exterior components, benefiting from enhanced mechanical properties and thermal stability. Examples include dashboards, instrument panels, and other trim components. The improved barrier properties and mechanical strength make CB/polystyrene nanocomposites suitable for packaging applications [19]. They can be used in food packaging, medical packaging, and other high-performance packaging solutions that require durability and protection against environmental factors. In construction, these nanocomposites can be used in insulation materials, panels, and other components that benefit from enhanced thermal stability, mechanical strength, and durability.

C. CB/Thermosetting Polymer Nanocomposites

Thermosetting polymers such as epoxy resins and phenolic resins play integral roles in high-performance applications across industries. The incorporation of carbon black (CB) into these matrices enhances the resulting nanocomposites' thermal stability, mechanical strength, and electrical conductivity, particularly benefiting sectors like aerospace, automotive, and electronics. Epoxy resins are valued for their exceptional mechanical strength, adhesion, and versatility, while phenolic resins excel in heat resistance, flame retardancy, and chemical durability. Both types of polymers are subjects of ongoing research aimed at further improving their performance and expanding their applications in sustainable materials solutions across diverse sectors [20].

The integration of CB into thermosetting polymers such as epoxy resins and phenolic resins yields nanocomposites with significantly enhanced properties, making them suitable for advanced applications. These advancements represent a notable step forward in composite materials technology, offering improved mechanical, thermal, electrical, and barrier properties. CB-reinforced epoxy resins and phenolic resins nanocomposites find wideranging applications due to their unique blend of properties and performance advantages.

CB, known for its high surface area, electrical conductivity, and reinforcing capabilities, is effectively combined with epoxy resins and phenolic resins to create nanocomposites with superior characteristics. Epoxy resins are widely used in structural applications, adhesives, and coatings owing to their robust mechanical strength and resistance to chemicals [21]. Phenolic resins, prized for their ability to withstand high temperatures and flames, are commonly employed in automotive, aerospace, and construction sectors. [22].The addition of CB enhances the tensile and flexural strength of both epoxy and phenolic resins. CB particles act as effective reinforcing agents, improving load-bearing capacity and resistance to mechanical stresses. CB reinforcement improves the impact resistance of nanocomposites, reducing the likelihood of crack propagation and enhancing toughness. It also improves the thermal stability of epoxy and phenolic resins by retarding the degradation process at elevated temperatures, thereby extending the operating temperature range and increasing durability in high-temperature environments. The high thermal conductivity of CB facilitates efficient heat dissipation throughout the nanocomposites, which is crucial for thermal management applications. CB forms a conductive network within the polymer matrix, imparting electrical conductivity to epoxy and phenolic resins [23]. This property is beneficial for applications requiring antistatic properties and electromagnetic interference (EMI) shielding. Furthermore, CB reinforcement improves the barrier properties of nanocomposites against gases and moisture. The presence of CB creates a tortuous path for gas and moisture molecules, reducing permeability through the resin matrix. The incorporation of CB enhances the abrasion and wear resistance of epoxy and phenolic resin nanocomposites, prolonging their lifespan in demanding applications. These nanocomposites exhibit improved resistance to chemicals and solvents, making them suitable for use in corrosive environments. CB reinforced epoxy and phenolic resin nanocomposites are utilized in automotive and aerospace industries for lightweight structural components, coatings, adhesives, and thermal insulation materials. Their enhanced mechanical strength, thermal stability, and flame retardancy make them ideal for critical applications. In the electronics industry, these nanocomposites find applications in circuit boards, electronic packaging, and EMI shielding components due to their electrical conductivity and mechanical robustness [24]. They are also employed in construction for high-performance coatings, adhesives, and sealants that require durability, chemical resistance, and thermal stability. In industrial settings, these materials are used for protective coatings, chemical-resistant linings, and components exposed to harsh environments. In conclusion, CB reinforced epoxy resins and phenolic resins nanocomposites offer a versatile and high-performance material solution across diverse industries [25]. Their enhanced mechanical, thermal, electrical, and barrier properties meet stringent application requirements, providing durability, reliability, and sustainability. Continued research and development in this field are expected to further optimize these nanocomposites, expanding their applications and advancing technological innovations in material science.

IV. CONCLUSION

Carbon black (CB) reinforced polymer nanocomposites represent a significant breakthrough in materials science, combining enhanced mechanical, electrical, and thermal properties. The selection of synthesis method significantly influences the performance of the resulting nanocomposite. Despite their applications across diverse industries, challenges remain, such as achieving uniform dispersion and strong interfacial adhesion. Future research efforts should prioritize the development of environmentally sustainable synthesis techniques, exploration of novel polymer matrices, and deeper insights into nanoscale interfacial interactions. Through ongoing innovation, CB reinforced polymer nanocomposites hold promise for transforming various applications, thereby driving advancements in technology and materials engineering.

V. REFERENCES

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